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A Report on the Use of a Conventional Wind Tunnel as a Multigas Facility

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Can you include the M=2.61 vertical traverse with CO2-air & Ar and He To?

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ABSTRACT

This report presents the results of utilizing the Jet Propulsion Laboratory 21-inch hypersonic and 20-inch supersonic wind tunnels as continuous flow multigas facilities. Carbon dioxide (CO_2) was used as the contaminating gas. Concentrations up to 70% by volume of CO_2 were obtained at Mach numbers 2.6 and 6.0. The techniques, special equipment and instrumentation used, and results are discussed. The results are of particular interest since aerodynamic testing in atmospheres approximating those of Venus and Mars appears to be feasible in facilities originally designed to use air as the working fluid.

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CONTENTS

FIGURES

I. INTRODUCTION

For some time increasing interest has been shown in testing spacecraft models in gases somewhat different from air. Consequently, it was desirable to determine if the two continuous-flow wind tunnels at the Jet Propulsion Laboratory (JPL) could be made to perform as multigas facilities.

A test program was initiated in the JPL 21-inch hypersonic wind tunnel (HWT) in which carbon dioxide (CO_2) was used as the contaminating gas. CO_2 was selected because of its stability ~~in~~ under the running conditions possible in the HWT, its non-toxicity, relative inexpensiveness, and availability. In addition, it is known that the atmospheres of both Venus and Mars, as compared with Earth, are rich in CO_2. If a ~~xxx~~ CO_2-rich environment could be established in the JPL wind tunnels, the utility of these tunnels could be extended, thus aiding JPL and NASA in their planetary exploration work.

This report will discuss the results of the initial program to establish the feasibility of operating the HWT as a multigas facility and the results of a subsequent study in the 20-inch supersonic wind tunnel (SWT).

II. THE PROBLEM

There were two separate problem areas. The first consisted of: 1) the introduction of a known quantity of CO_2 into the tunnel circuit; 2) the accurate measurement of the CO_2 concentration in the supply section of the tunnel; and, 3) the procedure for maintining, within acceptable limits, a constant concentration of CO_2 in the test section.

The nozzle contours normally used produce uniform flow distribution in the test section when air is the working fluid. Part two of the problem was to determine whether or not special contours would be required when a CO_2-rich atmosphere was the working fluid.

III. TEST TECHNIQUES

Three methods of introducing CO_2 into the tunnel circuit were tried. During the first test period in the HWT an attempt was made to introduce liquid CO_2 from a standard liquid-CO_2 container into the downstream end of the diffuser as shown in Figure 1. The CO_2 tended to freeze when expanded through the control valves thus plugging the supply line and resulting in intermittent flow of CO_2 to the diffuser.

For the second test period in the HWT it was decided to convert liquid CO_2 to the gaseous state before introduction into the diffuser. Control was greatly simplifed and very adequate but the necessity to generate steam for the liquid-to-gas converters was a nuisance.

Analysis of the control problem which prevailed during the first HWT test suggested that the liquid CO_2 should be metered by an orifice placed at the junction of the CO_2 supply line and the diffuser. Then the dry ice which is created during the expansion would sublime immediately in the diffuser air as shown in Figure 2. Figure 3 shows the arrangement of the CO_2 supply system which was developed and used successfully for the subsequent test in the SWT. In this sytem, the coarse metering is done by the orifices while the valves are used only for shut-offs or for small reductions in maximum flow to the orifices. Thus the pressure drop across the valves when there is flow through them is so small that no freezing occurs. A manifold of orifices such as shown in Figure 3 was chosen so that a wide variety of CO_2-air mixtures could be generated for several test-section Mach numbers. The Veejet nozzles which were obtained from System Spraying Co. produced a fan-shaped spray in a plane perpendicular to the air flow in the diffuser. This resulted in rapid evaporation and sublimation of the CO_2 as it entered the diffuser.

Throughout the program techniques were selected which took advantage of the conditions which normally exist in various parts of the tunnel circuit. For instance, CO_2 was introduced into the diffuser where the pressure is well below one atmosphere rather than in the supply pipe where the pressure is usually well above one atmosphere. This required the CO_2 to travel through the entire tunnel circuit before reaching the test section and therefore thorough mixing of the air and CO_2 was accomplished. In addition, it was convenient to take the sample of gas to be analyzed for CO_2 content from the tunnel ~~pipe~~ supply section.

The uniformity of the flow in the test section was checked by utilizing a semi-vertical traverse mechanism with two pitot probes.

IV. SPECIAL EQUIPMENT AND INSTRUMENTATION

The method of introduction of gaseous CO_2 into the HWT is shown in Figure 1. A large tank trailer with a maximum storage capacity of 19,000 pounds of liquid CO_2 under 250-300 psig pressure was used to supply liquid CO_2 to two liquid-to-gas converters. These converters were simple heat exchangers utilizing steam for the heat source. The gaseous CO_2 was then routed through either the flow meter for establishing low to moderate concentrations or through the 2-inch line for higher concentrations.

The temperature of the gaseous CO_2 was monitored throughout the test by a 5-millivolt Brown readout for the two thermocouples shown in Figure 1. The temperature of the gas was kept at about 140°F at the measurement points to inhibit the occurrence of a change of state at areas of expansion in the piping.

The SWT system required no special equipment in addition to the ~~Vujet~~ Veejet nozzles.

Liquid Carbonic, a Division of General Dynamics, supplied the CO_2, the steam generator, and the liquid-to-gas converters.

The concentration of CO_2 by volume was continually determined by using a Beckman Model 15a infrared analyzer. The output of the analyzer was read from a 5-millivolt Brown. A schematic drawing of the complete system is shown in Figure 4. A simple heat exchanger was used to keep the temperature of the gas sample between 70° and 100°F.

The analyzer was equipped with a reference cell which was calibrated for a selected range of CO_2 by volume in air. The reference cell used ~~during the HWT program~~ was calibrated for 0 to 15% CO_2 by volume. Cells for other ranges could be made but were not readily available ~~for the work in the HWT~~.

It was learned that adequate CO_2 percentage readings could be obtained up to 30% by volume using the 15% cell after experimentally extrapolating the calibration curve for the 15% cell. In addition, when operating with concentrations above 15%, samples of the gas mixture were collected periodically and carefully analyzed using chemistry laboratory techniques and equipment. ~~15 psia Statham pressure transducer~~ The pitot pressure is plumbed to a 15 psia Statham pressure transducer which supplies a ~~sign~~ signal that is continuously plotted during a traverse by a Moseley X-Y plotter. The X scale represents inches below the tunnel ceiling; the Y scale represents pressure units (cm Hg abs). More details concerning the gas analyzer are included in Reference 1.

The flow uniformity was checked by semi-vertical pitot-pressure traces. The HWT traverse has two pitot probes which ~~ixa~~ straddle the tunnel centerline and are 6 inches apart. When not in use, the probes are retracted flush with the tunnel ceiling. The traverse crosses the tunnel centerline approximately 15 inches upstream of the center of model rotation and moves in a plane inclined 15° to a plane perpendicular to the tunnel centerline. The SWT traverse is similar to the HWT instrument, but incorporates only one pitot probe.

V. RESULTS

The technique used to introduce gaseous CO_2 into the SWT circuit was most satisfactory and will also be used in the HWT. A sample operation is described in the appendix. Different levels of CO_2 concentration were established at nozzle contour settings corresponding to Mach numbers (M) for air of 1.8, 2.6, ~~5~~ 5, 6.5, and 7.2. At present the maximum Mach number at which testing can be done is near 7. The limiting factor is the maximum temperature available from the HWT heater. Air-CO_2 mixtures require much higher temperatures than air alone in order to avoid condensation effects. Experimental comparisons are shown in Table I. In Figure 5 a plot is presented which shows the effect on the pitot-pressure distribution of the onset of liquefaction for the M = ~~5~~ 2.6 and 5 nozzles ~~in the HWT~~. The pressure traces were obtained from semi-vertical traverses. Liquefaction effects at other concentrations of CO_2 were investigated, and indications were that the amount of CO_2 present seemed to have only a minor effect on the temperature at which liquefaction effects appeared. The explanation of ~~the~~ this apparent anomaly is not ~~xxxxxx~~ obvious and will require more analytic and experimental work. It ~~is~~ was noted, however, that as the supply pressure ~~for~~ CO_2-air mixtures was reduced, liquefaction occurred at a lower temperature which is consistent with liquefaction effects for "pure" air.

The maximum concentraton of CO_2 which was run was 70% by volume. This was achieved in the M = 6 nozzle for a supply pressure of 900 cm Hg abs and a supply temperature of 1200°F. To maintain this concentration, it was necessary to continuously put 2 to 3 pounds of CO_2 per second into the tunnel to make up for leakage of air into the below-atmosphere part of the HWT circuit.

The ability to accurately measure the concentration of CO_2 in the tunnel was demonstrated. At concentrations from 0 to 15%, the analyzer readout gave readings which were accurate to ± 0.1% CO_2 by volume*. Accuracy decreased somewhat at higher concentrations, but significant test results could still be obtained.

Attempts were also made to establish and hold a desired concentration. After gaining a little experience, concentrations could be quickly established to within 0.5% and held to ± 0.2% of that desired.

no ¶

CO_2 was easily purged from the tunnel circuit by evacuating the tunnel to a low pressure after bypassing the tunnel flow at the end of a run. This purging was accomplished in from 2 to 3 minutes.

When liquefaction was avoided, the nozzle contours normally used for "pure" air were adequate for air-CO_2 mixtures for CO_2 concentrations up to 70% by volume.

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* The instrument was previously calibrated to this accuracy by using known mixtures of CO_2 and air.

VI.. FUTURE WORK

The effect of CO_2-air mixtures on aerodynamic ooefficients will be studied in the SWT ~~xx~~ and HWT. Some of this work has recently been performed and will be discussed in a subsequent report.

Better control of the introduction of CO_2 into the tunnel circuit is certainly possible. A servo control can be developed which would automatically hold the concentration of CO_2 relatively constant.

Table I

Mach No.*	Total Pressure (cm Hg abs.)	Approximate CO_2 Concentration (% by Vol. in air)	Total Temp. Below Which Liquafaction Effects Appear (^{o}F) For CO_2 Air Mixtures	For Air Alone
5.07	515	11.6	775	150
6.49	200	15.2	950	300
6.52	200	10.0	950	300
6.56	800	5.4	1200	420
6.56	800	10.5	1200	420
7.13	200	14.4	1200	380

* Mach number was calculated using standard air tables. No attempt was made to determine the "real" Mach number at this time.

APPENDIX

Sample Operatinn in the SWT

For the M = 1.8 nozzle contour and a supply pressure of 80 cm Hg abs and a supply temperature of 115°F, the mass flow thro ugh the tunnel is 85 lb/sec. To test in a mixture of 20% CO_2 and 80% air by volume one does the following after starting the tunnel and reaching the above supply conditions (refer to Figure 3).

1. Fully open the 3/8-inch ball valve which raises the CO_2 content of the tunnel "air" to 20% by volume in about 4 min.; CO_2 mass flow is 3 lb/sec.

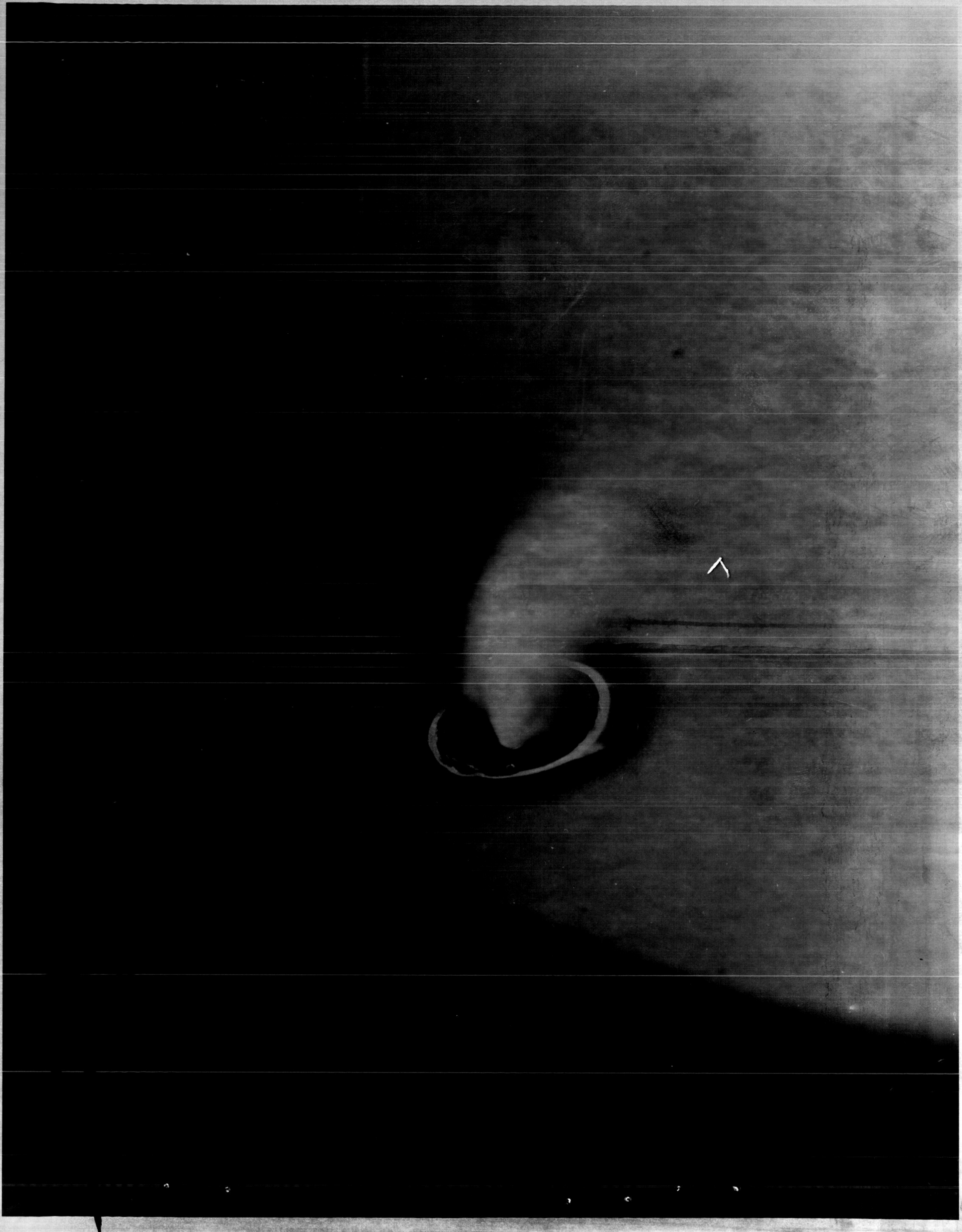
2. Close the ball valve and maintain the 20% CO_2 concentration with the 1/4-inch globe valves and ~~Muljetxnozzlesx~~ Vecjet nozzles; CO_2 mass flow is 1 lb/sec.

3.

REFERENCE

1. Beckman Instruction Manual for the L/B Infrared Analyzer Model 15a.

FIG 1



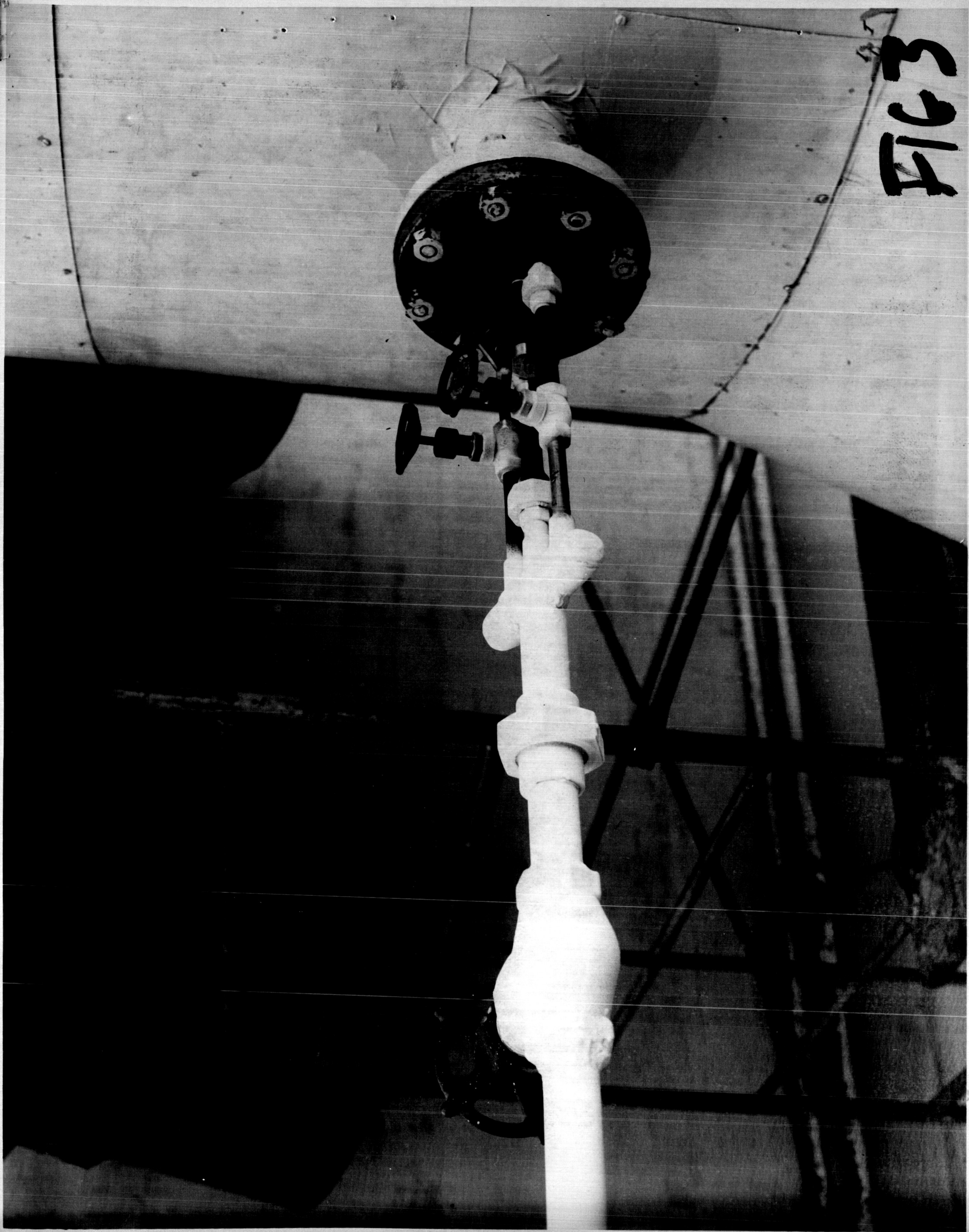
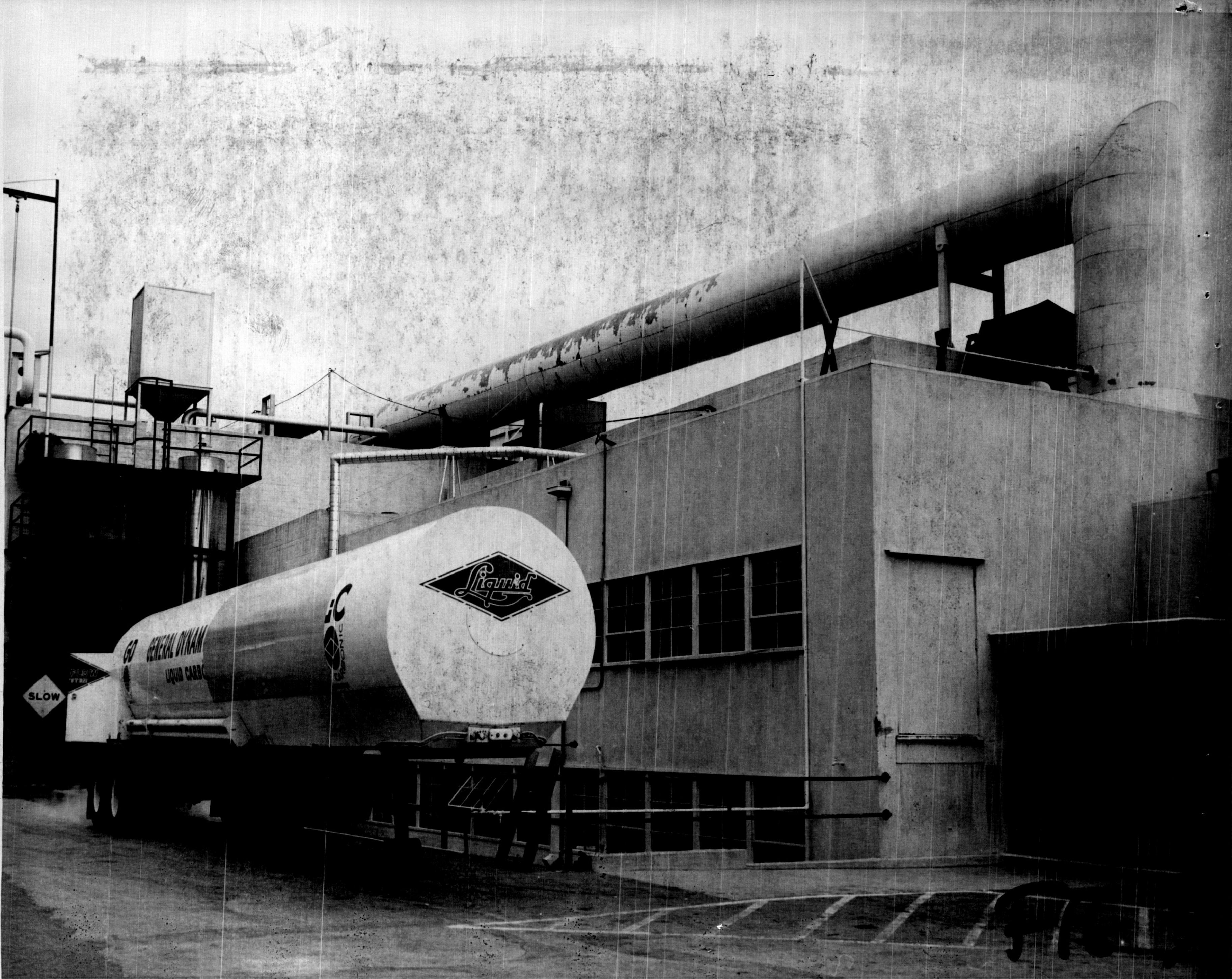
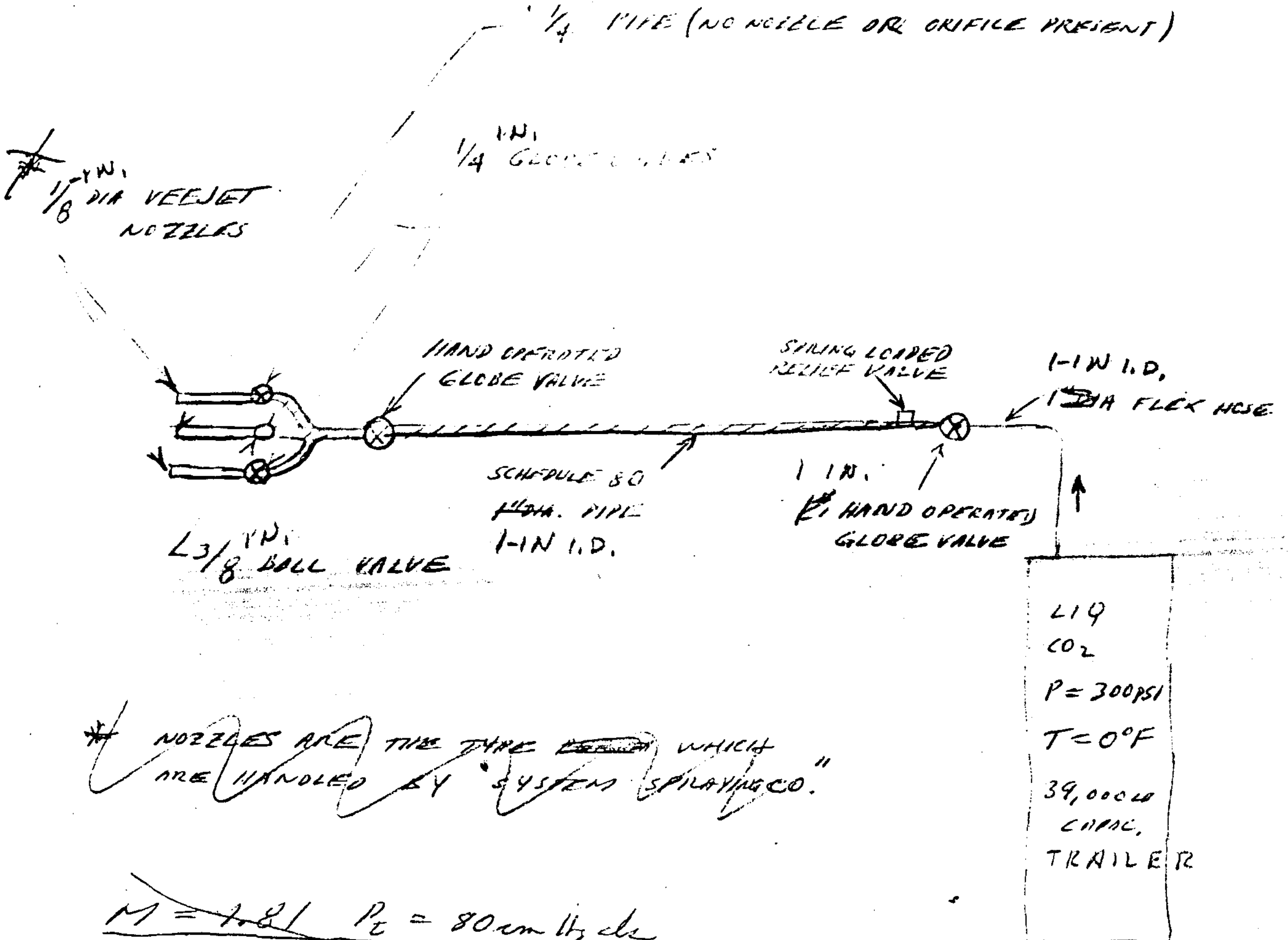


FIG 3


Liquid
GENERAL DYNAM
LIQUID CARBO
GD
SLOW

GH
4/16/63
20-C96

SCHEMATIC FOR CO_2 FEASIBILITY
STUDY IN 20-IN. S.W.T.

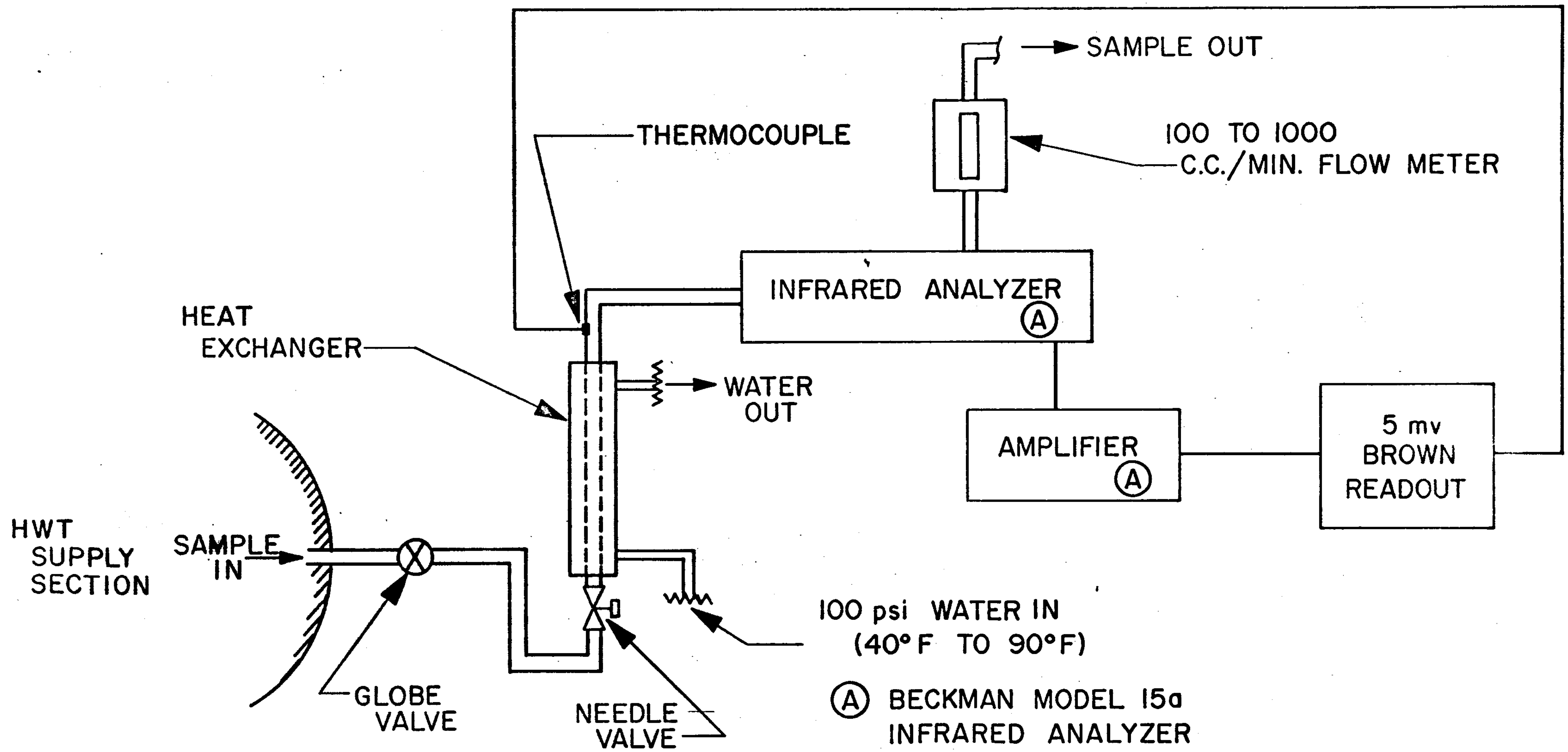


* NOZZLES ARE THE TYPE WHICH ARE HANDLED BY "SYSTEM SPRAYING CO."

~~$M = 1.81$ $P_t = 80$ cm Hg abs~~

TO OPERATE, 3/8" BALL VALVE IS FULLY OPENED WHICH BRINGS THE CO_2 CONCENTRATION TO 20% IN APPROX. 4 MIN., THEN CLOSE VALVE, BLEED IN SMALL AMOUNT THROUGH 1/8" DIA. VEEJET NOZZLE.

No. Figure 3. Schematic



CO_2 SAMPLING SYSTEM

FIG 4

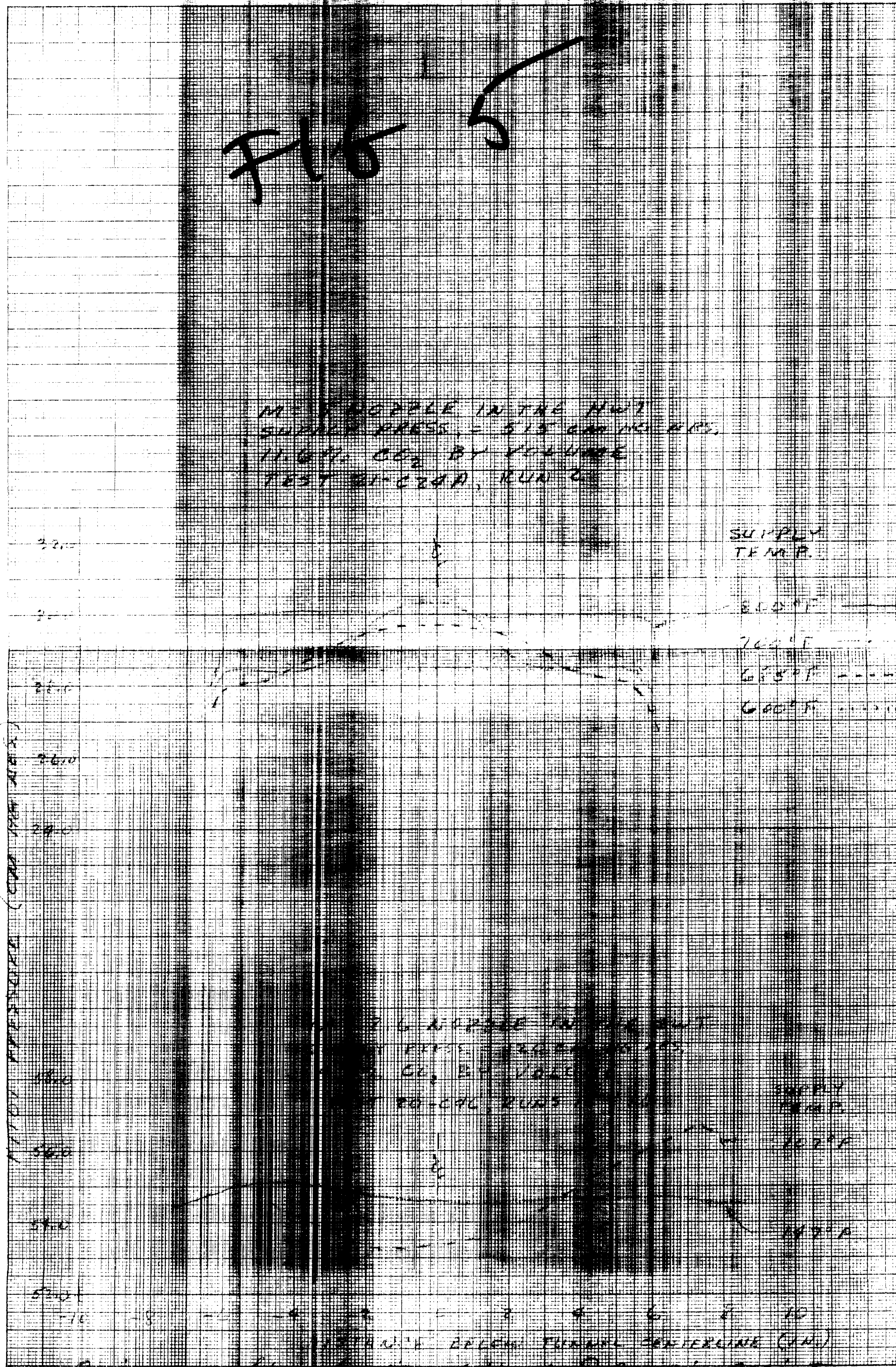


Figure 5 Interference Effects of CO_2-Air Mixture on Pitot-Pressure Distribution